

## Two new compounds, $\beta$ -ScTe and $Y_3Au_2$ , and a reassessment of $Y_2Au$

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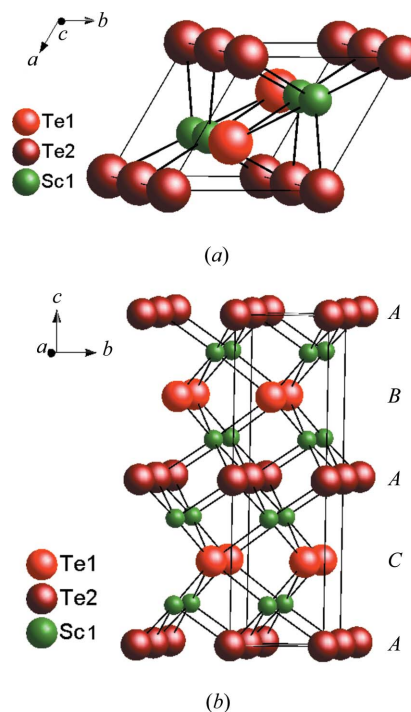
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Two new compounds,  $\beta$ -ScTe (scandium telluride) and  $Y_3Au_2$  (triytrium digold), have been synthesized by high-temperature solid-state techniques and their crystal structures, along with that of  $Y_2Au$  (ditytrium gold), have been refined by single-crystal X-ray diffraction methods.  $\beta$ -ScTe is a superstructure of ScTe (NiAs-type), featuring double hexagonal close-packed layers of Te atoms with the octahedral cavities filled by Sc atoms.  $Y_3Au_2$  displays a  $U_3Si_2$ -type structure and is built from  $Au_2$ -centered bitrigonal prisms and centered cubes of Y atoms. The structure of  $Y_2Au$  is better described as an inverse  $PbCl_2$ -type structure rather than a  $Co_2Si$ -type.

### Comment

Research within solid-state chemistry on ternary and polynary compounds has attracted much attention during the past few decades because of their interesting structures, bonding and physical properties. Knowledge of binary compounds can provide significant references during the exploratory synthesis of polynary compounds, and their identification is therefore valued. Although many binary combinations of elements are covered by binary phase diagrams and the crystal databases, some have been missed because of the limitations of the earlier experiments. For instance, only two compounds, *viz.* ScTe and  $Sc_2Te_3$ , were in the Sc–Te phase diagram reported in 1990 (Okamoto, 1990), missing several examples,  $Sc_2Te$ ,  $Sc_8Te_3$  and  $Sc_9Te_2$ , which were discovered later during the study of ternary systems (Maggard & Corbett, 1997, 1998, 2000). We discovered that the structure of  $\beta$ -ScTe was still missing, with an inverse  $Li_2O_2$ -type structure, a double hexagonal close-packed (dhcp) version of ScTe (NiAs-type) (Menkov *et al.*, 1961).  $Y_3Au_2$  was missed in the investigation of the Y–Au phase diagram (Saccone *et al.*, 1997), whereas  $Y_2Au$  was identified as  $Co_2Si$ -type (Yakinthos *et al.*, 1978) from lattice parameters only; no refinement of powder diffraction intensities was carried out. The crystal structures of these three phases are described here.

$\beta$ -ScTe is presumably a high-temperature phase with an inverse  $Li_2O_2$ -type structure. A view of the unit cell approximately along [001] is shown in Fig. 1(a) and a section

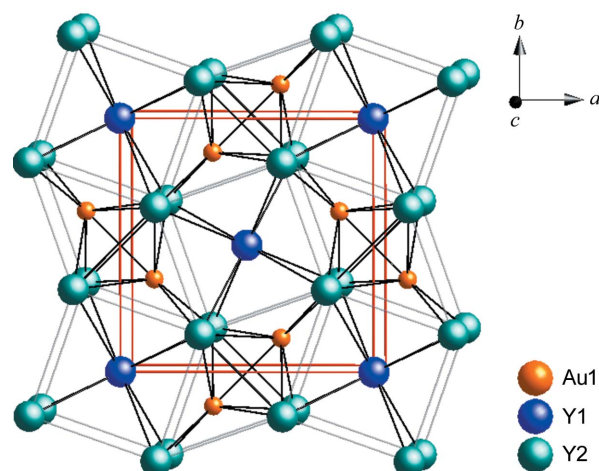


**Figure 1**

(a) A view of one unit cell, approximately along [001]. (b) A section of  $\beta$ -ScTe, approximately along [100].

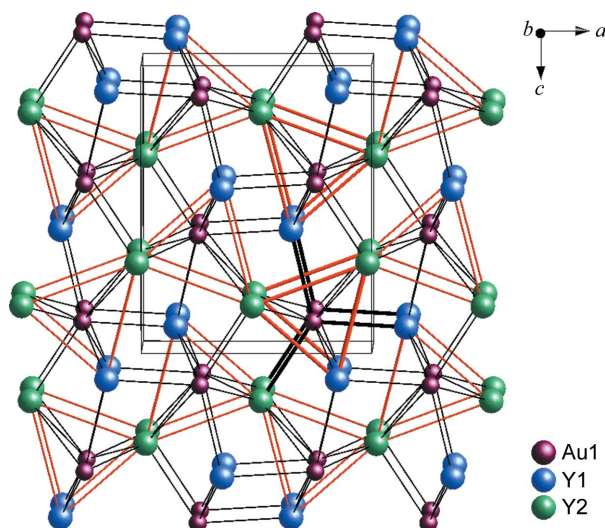
approximately along [100] is shown in Fig. 1(b), in which Te atoms form hexagonal close-packed (hcp) layers with  $ABAC\dots$  stacking, leaving all nominal octahedral cavities filled by Sc atoms. Note that ScTe (NiAs-type) crystallizes in the same space group,  $P6_3/mmc$ , with  $a = 4.130(5)$  Å and  $c = 6.749(5)$  Å, and contains simple hcp Te atoms of  $ABAB\dots$  ordering with Sc atoms occupying the octahedral cavities (Menkov *et al.*, 1961). Therefore,  $\beta$ -ScTe is a stacking variant of ScTe, with a  $c$  axis twice as large.

$Y_3Au_2$  crystallizes in the  $U_3Si_2$ -type structure in the space group  $P4/mbm$  (No. 127). An approximately [001] projection along the short  $3.907(3)$  Å  $c$  axis is shown in Fig. 2. The basic building units are an  $Au_2$ -centered bitrigonal prism (BTP) of



**Figure 2**

A projection of  $Y_3Au_2$ , approximately along [001].



**Figure 3**  
A view of  $Y_2Au$ , approximately along  $[001]$ .

$Y_2$  atoms and a  $Y_1$ -centered  $Y_2$  cube. The  $Au_2$  unit is an unusual dimer  $3.0539(19)$  Å long. The two-dimensional motif is created in such a way that each  $Y_2$  cube, with the centered  $Y_1$  atom on a fourfold axis, interconnects with four identical units through shared  $Y_2$ – $Y_2$  edges, leaving the cavities filled by the BTPs. The centered  $Y_1$  and  $Au_1$  atoms lie on a mirror plane at  $c = 0$ . The two-dimensional motif repeats along  $c$  to form a three-dimensional network, sharing bitriangular and square faces.

The structure of  $Y_2Au$  in the space group  $Pnma$  projected along  $[010]$  is shown in Fig. 3, in which each  $Au$  atom forms the center of a trigonal prism (TP) of  $Y_1$  or  $Y_2$  atoms that shares trigonal faces with identical units to generate an infinite one-dimensional column along  $b$ . The TPs interconnect with shared  $Y_2$ – $Y_2$  edges and lead to the formation of puckered sheets along  $a$ . Adjoining sheets stack along  $c$ , with displacements of  $b/2$ , to create a three-dimensional structural network. As a result, each TP is tricapped by two  $Y_1$  atoms and one  $Y_2$  atom so that each  $Au$  atom has nine neighbors.  $Y_2Au$  is better described as an inverse  $PbCl_2$ -type structure rather than the earlier reported  $Co_2Si$ -type, in which the  $Si$  atom would have ten neighbors (Flahaut & Thévet, 1980; Liu & Corbett, 2006).

## Experimental

The formation of all three phases was in fact first noted in powder pattern data from nearby ternary systems.  $\beta$ - $ScTe$  was synthesized from a mixture of  $Sc$  and  $Sc_2Te_3$ . The latter was obtained from a pre-reaction of  $Sc$  and  $Te$  in a 2:3 ratio, sealed in a silica tube under vacuum, heated at 723 K for 12 h and then at 1173 K for 72 h.  $Y_3Au_2$  and  $Y_2Au$  were prepared from mixtures of high-purity elements. The three combinations were each pelletized and arc-melted under an argon atmosphere in a glove-box. The pellets were then sealed in tantalum containers and annealed in a graphite-heated vacuum furnace at 1573 K for 200 h for  $\beta$ - $ScTe$ , and at 1323 K for one week for  $Y_3Au_2$  and  $Y_2Au$ . The products were crushed with the aid of an agate mortar into small crystals that were suitable for single-crystal X-ray diffraction.

## ScTe

### Crystal data

$ScTe$   
 $M_r = 172.56$   
Hexagonal,  $P6_3/mmc$   
 $a = 4.0969(6)$  Å  
 $c = 13.602(3)$  Å  
 $V = 197.71(6)$  Å<sup>3</sup>

$Z = 4$   
Mo  $K\alpha$  radiation  
 $\mu = 17.64$  mm<sup>-1</sup>  
 $T = 293$  K  
 $0.15 \times 0.08 \times 0.07$  mm

### Data collection

Bruker SMART CCD area-detector diffractometer  
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{min} = 0.177$ ,  $T_{max} = 0.372$

1274 measured reflections  
92 independent reflections  
90 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.020$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.081$   
 $S = 1.57$   
92 reflections

8 parameters  
 $\Delta\rho_{max} = 1.21$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -1.65$  e Å<sup>-3</sup>

## $Y_3Au_2$

### Crystal data

$Y_3Au_2$   
 $M_r = 660.66$   
Tetragonal,  $P4/mbm$   
 $a = 8.059(3)$  Å  
 $c = 3.907(3)$  Å  
 $V = 253.7(3)$  Å<sup>3</sup>

$Z = 2$   
Mo  $K\alpha$  radiation  
 $\mu = 91.35$  mm<sup>-1</sup>  
 $T = 293$  K  
 $0.05 \times 0.04 \times 0.02$  mm

### Data collection

Bruker SMART CCD area-detector diffractometer  
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{min} = 0.092$ ,  $T_{max} = 0.262$

1866 measured reflections  
194 independent reflections  
156 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.093$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.030$   
 $wR(F^2) = 0.069$   
 $S = 0.93$   
194 reflections

11 parameters  
 $\Delta\rho_{max} = 1.46$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -1.40$  e Å<sup>-3</sup>

## $Y_2Au$

### Crystal data

$Y_2Au$   
 $M_r = 374.79$   
Orthorhombic,  $Pnma$   
 $a = 7.115(3)$  Å  
 $b = 4.933(2)$  Å  
 $c = 8.908(4)$  Å

$V = 312.7(2)$  Å<sup>3</sup>  
 $Z = 4$   
Mo  $K\alpha$  radiation  
 $\mu = 83.29$  mm<sup>-1</sup>  
 $T = 293$  K  
 $0.12 \times 0.08 \times 0.04$  mm

### Data collection

Bruker SMART CCD area-detector diffractometer  
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{min} = 0.036$ ,  $T_{max} = 0.135$

1619 measured reflections  
345 independent reflections  
299 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.039$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.026$   
 $wR(F^2) = 0.058$   
 $S = 1.07$   
345 reflections

20 parameters  
 $\Delta\rho_{max} = 1.76$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -2.14$  e Å<sup>-3</sup>

For all three title compounds, data collection: *SMART* (Bruker, 2002); cell refinement: *SAINT-Plus* (Bruker, 2003); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *DIAMOND* (Brandenburg, 2000); software used to prepare material for publication: *SHELXTL*.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: FN3086). Services for accessing these data are described at the back of the journal.

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